



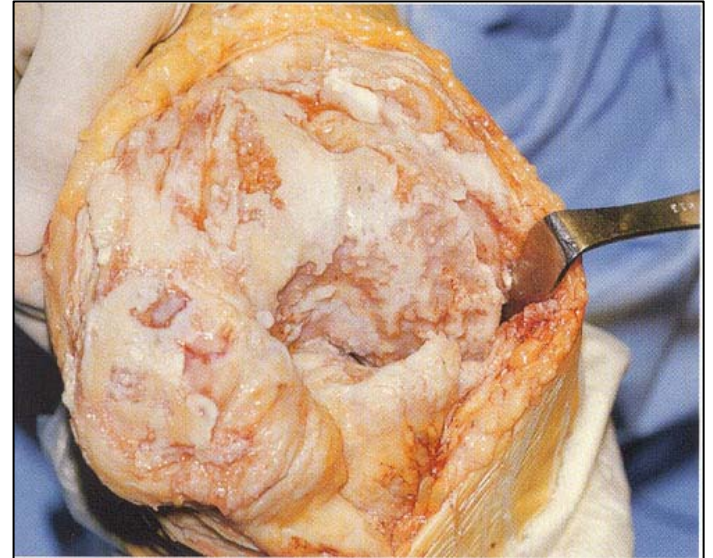
Knee Replacement Therapy

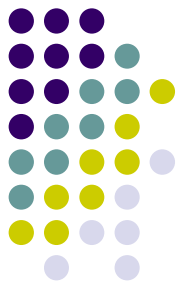
Primary Problem:

- Damaged cartilage leads to various forms of arthritis
- Osteoarthritis: 20.7 million Americans

Symptoms:

- hard, bony swelling of the joints
- gritty feeling
- Immobility





Introduction - Background

Solution: *Total Knee Replacement (TKR)*

- Nearly 250,000 Americans receive knee implants each year

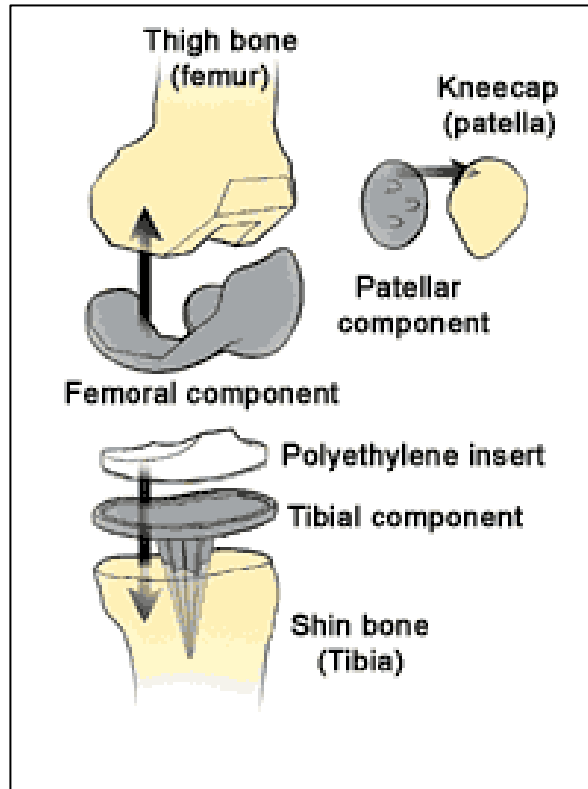
Results:

- Stops or greatly reduces joint pain
- Improves the strength of the leg
- Increases quality of life and comfort





Current TKR Design - Assembly



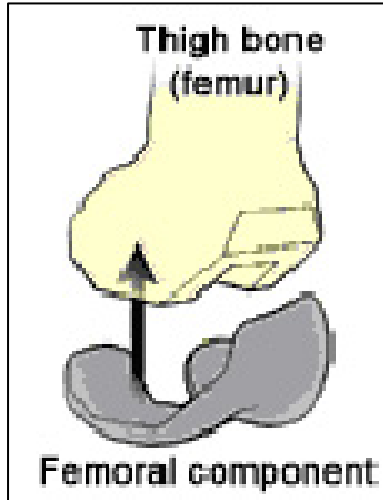
Four Primary Components:

1. Femoral Component
3. Plastic Insert

2. Tibial Component
4. Patellar Component



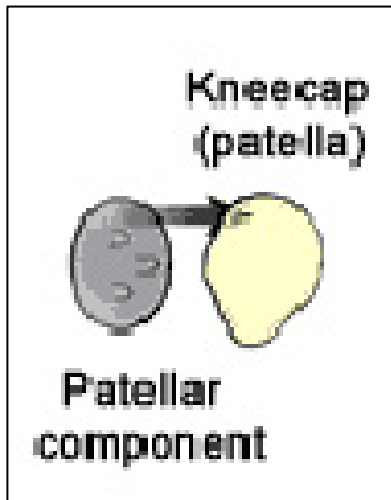
Current TKR Design - Components



Femoral Component

Materials: Cobalt-chromium-molybdenum
Ti-6Al-4V ELI Titanium Alloy

Interface: Press fit, biological fixation, PMMA



Patellar Component

Materials: Polyethylene
Cobalt-chromium-molybdenum (Ti Alloy)

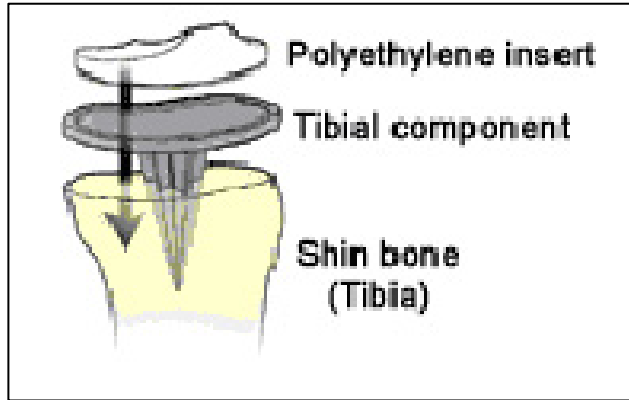
Interface: Press fit,
biological fixation
PMMA

*Modular or singular design





Current TKR Design - Components



Tibial Component

Materials: Cobalt-chromium-molybdenum (cast)
Ti-6Al-4V ELI Titanium Alloy

Interface: Press Fit, Biological Fixation, PMMA

Plastic Insert

Materials: Polyethylene

Interface: Press Fit



Current TKR Design - Problems

#1 Polyethylene “The Weak Link”

- Articulation wear produces particulates
- Leading to osteolysis and bone resorption at the implant interface.
- loosening and eventual malfunction of the implant will occur.



Worn UHMWPE cup
of total hip replacement.



Worn UHMWPE component of
total knee replacement.

Figure 2.3 Examples of worn UHMWPE components for total hip and knee replacements.



Current TKR Design - Problems

#1 Polyethylene “The Weak Link”

- Articulation wear produces particulates
- Leading to osteolysis and bone resorption at the implant interface.
- loosening and eventual malfunction of the implant will occur.

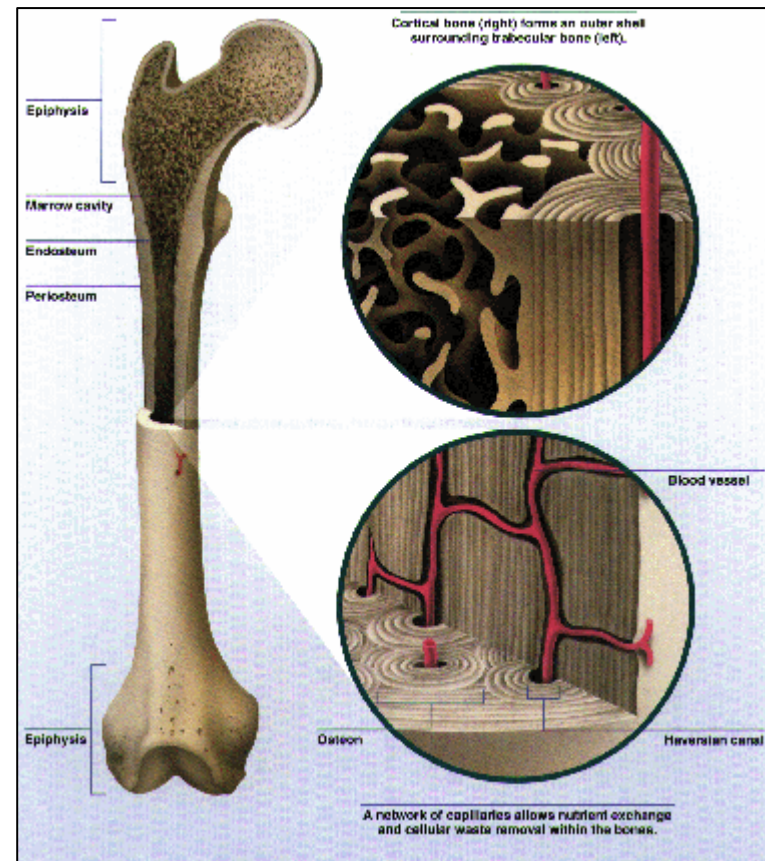
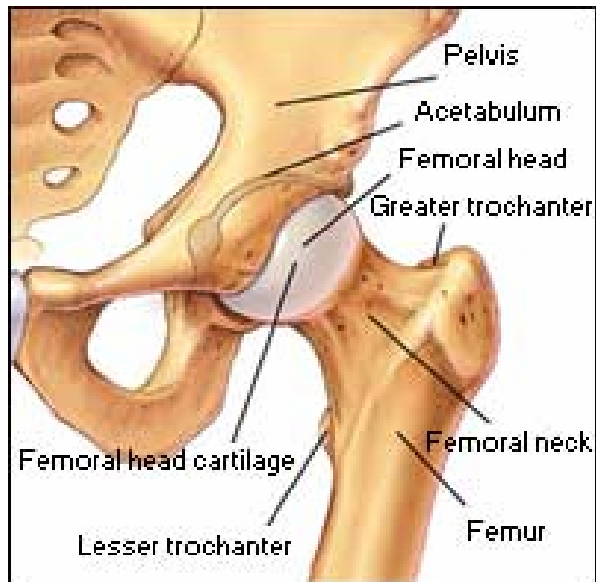
#2 Metal-Bone Interface

- Stress-shielding leads to bone degeneration



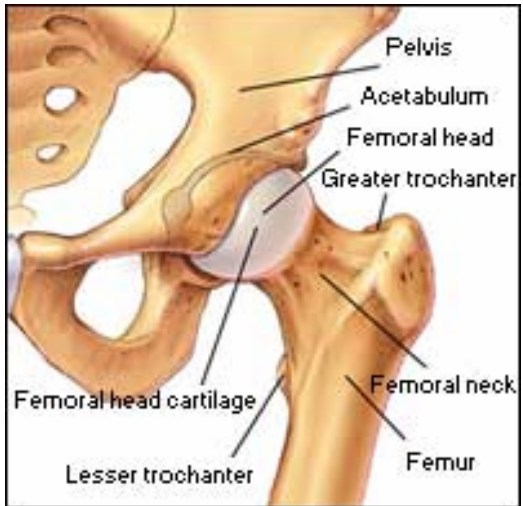
Average lifespan of 10-20 years

Total Artificial Hip



The surgical procedure

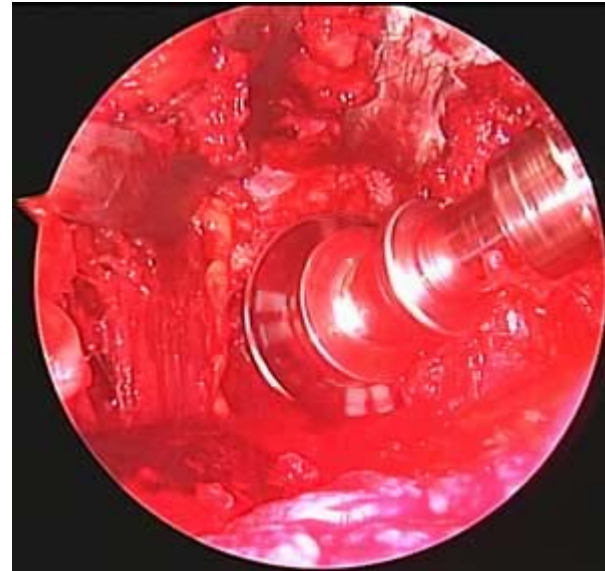




The surgical procedure

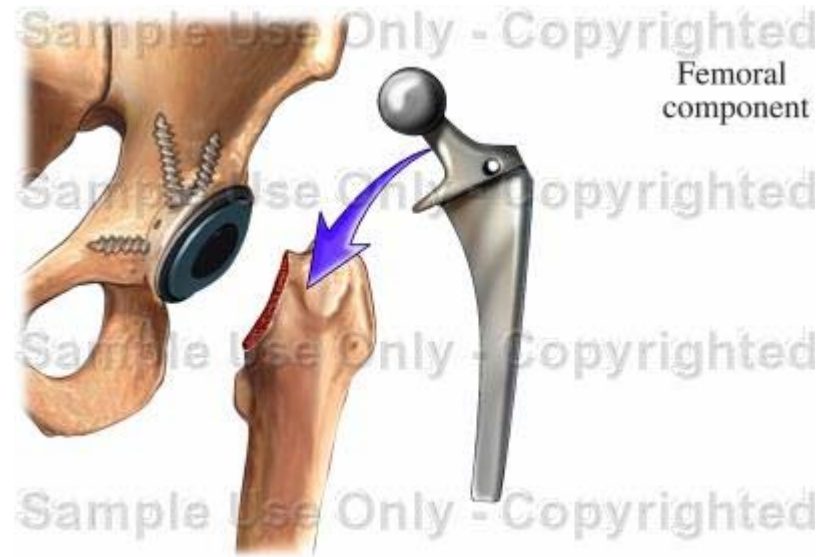
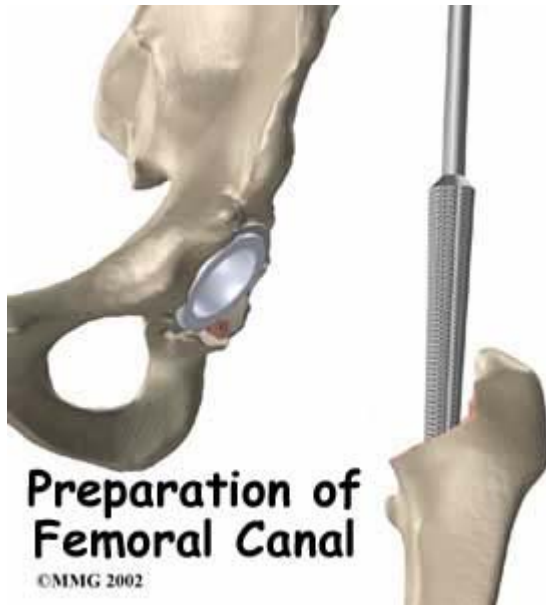


Mini Incision Hip Replacement



Preparation of the acetabulum is obtained using specially designed acetabular reamers (like cheese graters).





Bone Cement

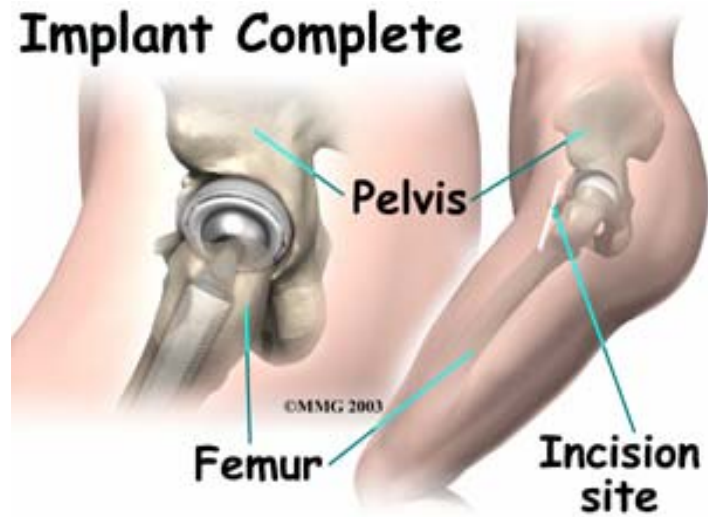


MIXEVAC® III



Revolution™

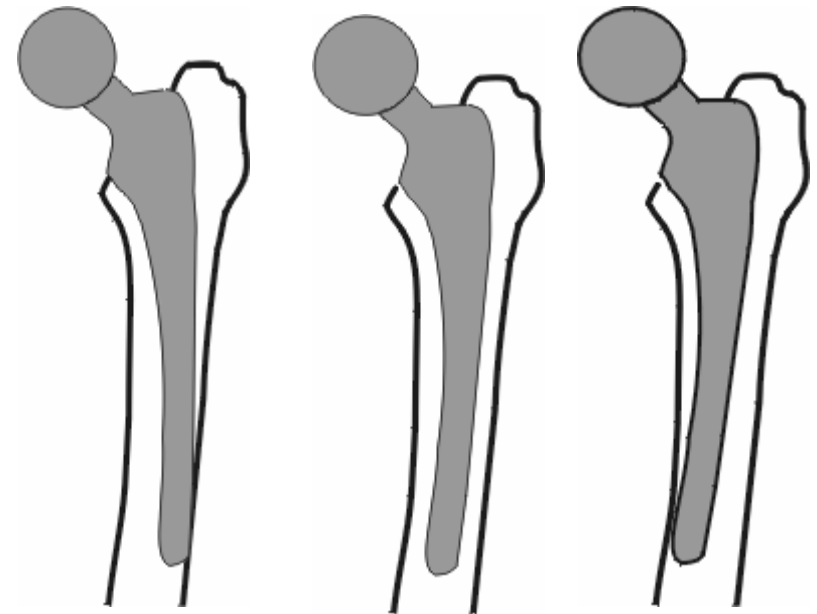
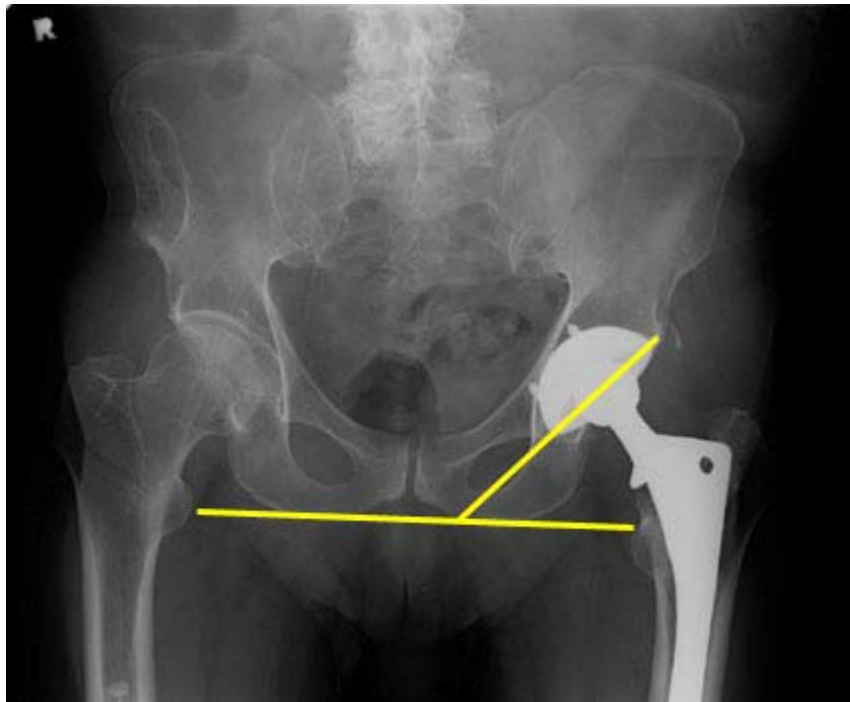
- a mixture of PMMA particles, the liquid monomer MMA (methyl methacrylate), a radio-opaque barium salt, and initiator (organic peroxide) to start the polymerization reaction of the MMA to PMMA.



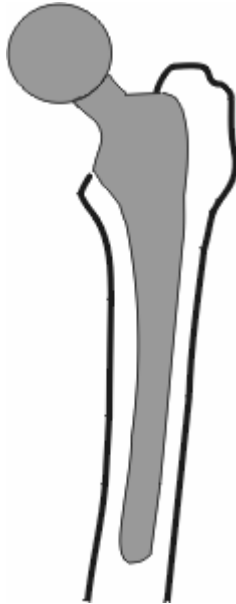
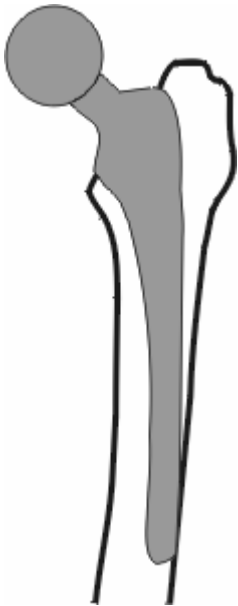
Mini-Incision total hip Replacement



HIP ARTHROPLASTY: Alignment and Stem Position



Stem Position



A Number of Components to Choose from:





Polyethylene and porous acetabular cups, back and front views



Metal acetabular cup with polyethylene liner, disassembled and assembled



Metal acetabular cup with metal liner, disassembled and assembled



Metal acetabular cup with polyethylene liner disassembled, with various ceramic and metallic femoral component heads



Acetabular cups—ceramic, metal and polyethylene, and porous ingrowth



Porous ingrowth femoral stem with various metal and ceramic heads



Porous ingrowth total hip replacement with polyethylene cup line

Major factors causing failure of total joint replacements include:



1. Infection during orthopedic surgery;
2. Fracture of the implants;
3. Fixation problem of the implants;
4. Wear of the implant materials; and,
5. **Osteolysis induced by wear particles.**



Worn UHMWPE cup
of total hip replacement.



Worn UHMWPE component of
total knee replacement.

Figure 2.3 Examples of worn UHMWPE components for total hip and knee replacements.

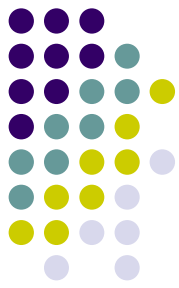


Table 2.2 Summary of retrieval UHMWPE wear particle studies in the literature.

Reference	Sample source	Particle isolation	Material analysis	Particle analysis
Wolfarth et al. [35] <i>JBMR</i> , vol. 34, pp.57-61 (1997) *FDA	Synovial fluid, tissues (TKR)	NaOH digestion, ultra-centrifugation, water and ethanol centrifugation	Micro-Ramman spectroscopy	SEM The particles from synovial fluid and tissue are not much different. Smooth, granular shapes, thin, irregularly-shaped, shaving like particles.
Campbell et al. [36] <i>JBMR</i> , vol. 29, pp. 127-131 (1995) *LA orthopedic hospital	Tissues (THR)	NaOH digestion, density gradient ultra-centrifugation, sucrose density gradient centrifugation	FTIR, DSC	SEM * Rounded (0.3 μm) * Elongated (1.9 μm) * Fibrils with rounded heads at one end, thin taper at the other
Shanbhag et al. [37] <i>JBJS</i> , vol. 76-B, pp. 60-67 (1994) *Rush-Presbyterian-St Luke's medical center	Tissues, interfacial membranes (THR)	KOH digestion, centrifugation, hexane-ethanol fractions	FTIR, EDX	SEM * Spheroids: 0.1 μm to 0.2 μm * Fibrils: 0.2 μm to 0.3 μm wide, up to 10 μm long. * 92 % particles are smaller than 1 μm .
Hailey et al. [38] <i>PIME</i> , vol. 210, pp. 3-10 *U. of Leeds, UK	Tissue (THR)	KOH digestion, centrifugation, 0.2 μm filtration	N/A	SEM 0.3 μm up to 3 μm
Lee et al. [39] <i>JBJS</i> , vol. 74-B, pp. 380-384 (1992) *Hospital for special surgery, NY	Tissues (THR)	Digestion in tissue solubiliser, centrifugation	N/A	Polarized light microscopy Short dimension 2 μm to 4 μm , long dimension 8 μm to 13 μm .
Maloney et al. [40] <i>JBJS</i> , vol. 77-A, pp. 1301-1310 (1995) *Stanford U.	Tissues (THR)	Papain digestion	X-ray	SEM, Coulter multisizer II * Fixed cup: 0.4 μm * Bipolar: 0.7 μm
Schmalzried et al. [41] <i>JBMR</i> , vol. 38, pp. 203-210 (1997) *LA orthopedic hospital	Tissues (THR, TKR)	NaOH digestion, sucrose ultra-centrifugation, isopropanol density gradient ultra-centrifugation	FTIR, DSC	SEM * Granules: sub-micron * Beads: 1 μm to 2 μm * Fibrils: up to 5 μm * Shreds: 10 μm to 20 μm long, up to several microns wide.
Hirakawa et al. [42] <i>JBMR</i> , vol. 31, pp. 257-263 (1996) *The Cleveland clinic foundation	Tissues (THR, TKR)	70 % Nitric acid digestion, centrifugation, 0.4 μm filtration	EDX	SEM, Coulter multisizer II * <10 μm particles: 0.72 μm (hip), 0.74 (knee) * >10 μm particles: 81 μm (hip), 283 μm (knee)

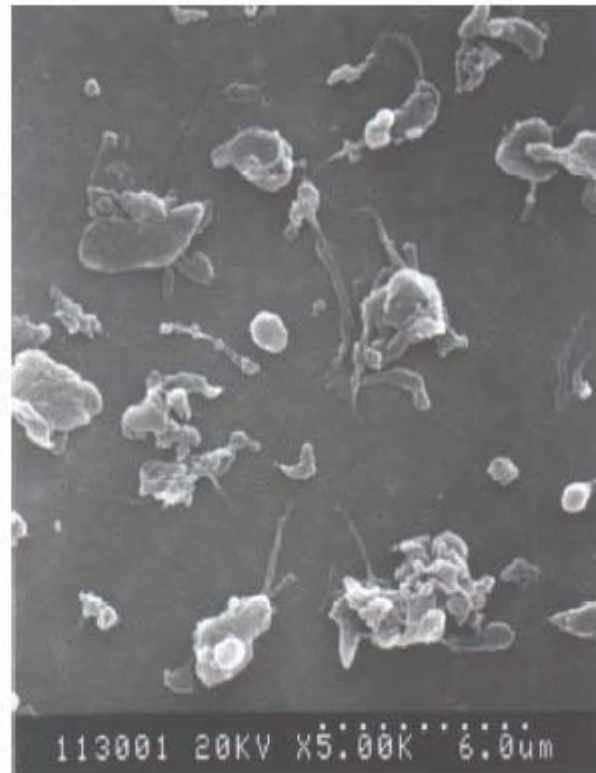


Figure 2.4 Examples of UHMWPE particles retrieved from total knee replacement.

Wear particle induced osteolysis



Figure 2.5 Osteolysis taking place around the total hip replacements which is detected by X-ray diagnosis.

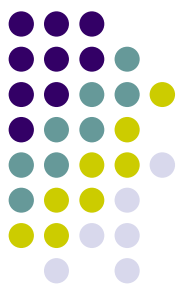
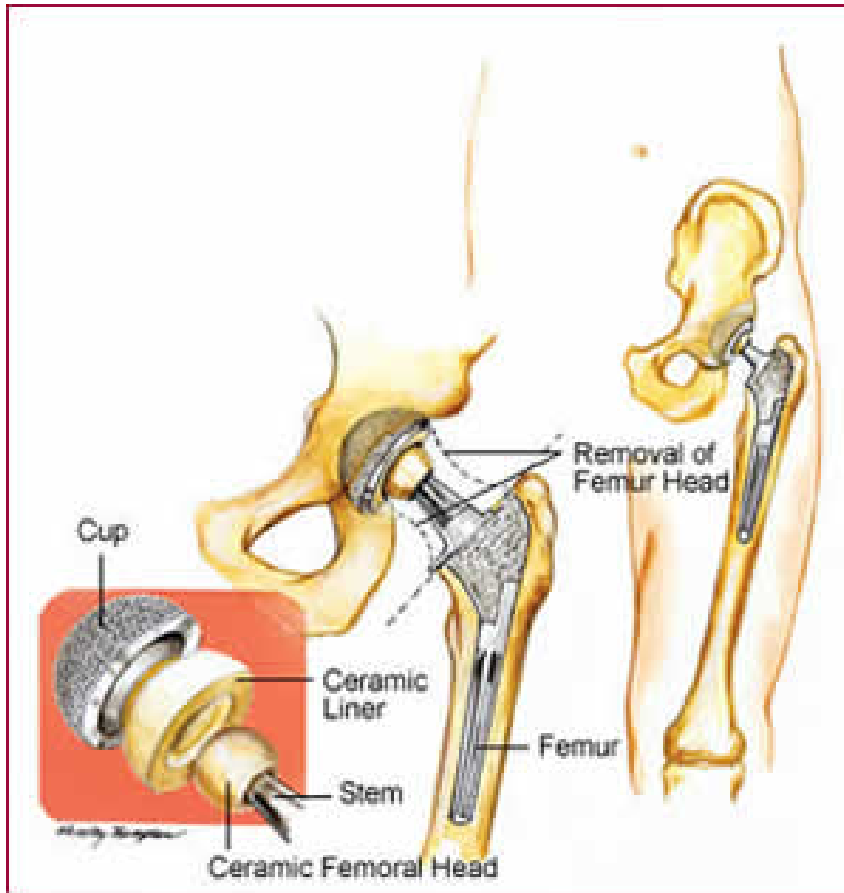


Table 2.3 Definition of biological terms related to osteolysis

Terms	Definition
Osteolysis	Dissolution of bone through disease, commonly due to infection or by loss of blood supply to the bone.
Necrosis	The death of some of all the cells in an organ or tissue, caused by disease, physical or chemical injury, or interference with the blood supply.
Phagocytosis	The engulfment and digestion of bacteria and other foreign particles by a cell.
Macrophage	A large scavenger cell (a phagocyte) present in connective tissue and many major organs and tissues, including the bone marrow, spleen, lymph nodes, liver, and the central nervous system.
Osteoblast	A cell, originating in the mesoderm of the embryo that is responsible for the formation of bone.
Osteoclast	A large multinucleate cell that resorbs calcified bone. Osteoclasts are only found when bone is being resorbed and may be seen in small depressions on the bone surface.
Cytokine	Soluble mediators secreted by macrophages, controlling many critical interactions among cells of the immune system.
IL-1β, IL-6	Cytokine, interleukins mediators
TNF-α	Cytokine, tumor necrosis factor α
GM-CSF	Cytokine, colony-stimulating factors
PGE-2	Cytokine, prostaglandins

Hip Implant



Osteointegration

The leap



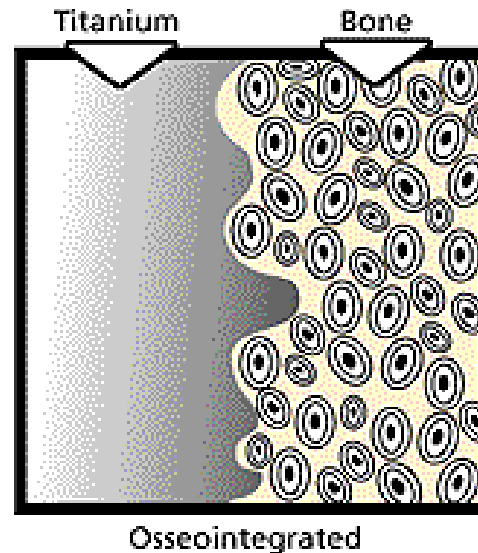
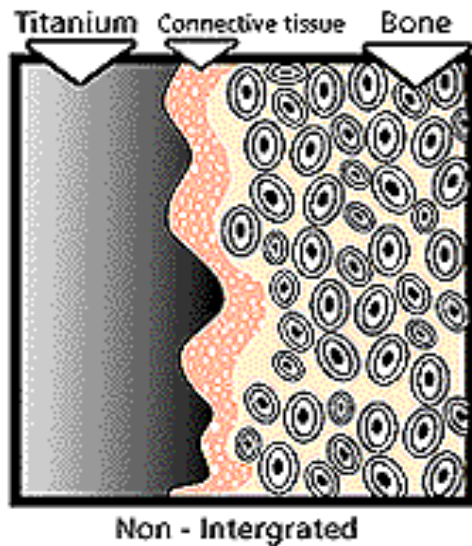
- 1952 - Per Ingvar Branemark,
- Discovered the titanium screw.
- Introduced the concept of Osseointegration



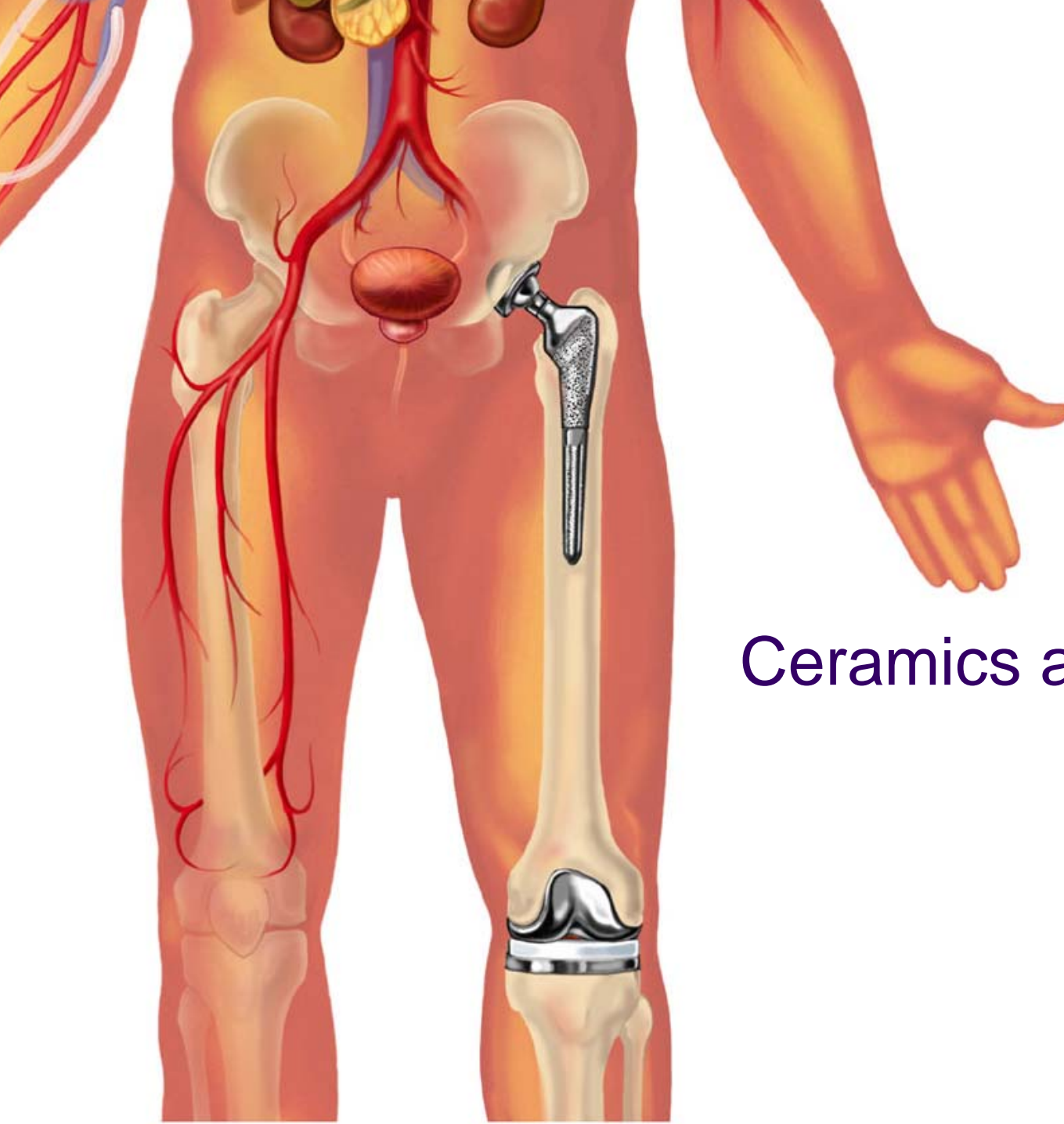
All existing designs based on
Branemark Titanium Screw



Osseointegration – The Divine Mantra



A fixture is osseointegrated if it provides a stable and apparently immobile support of prosthesis under functional loads, without pain, inflammation, or loosening.



Ceramics and Glasses

Definitions



Ceramic: an inorganic, nonmetallic, typically crystalline solid that is prepared from powdered materials and is fabricated into products through the application of heat.

- Most ceramics are made up of two or more elements.
- Inorganic compounds that contain metallic and non-metallic elements, for which inter-atomic bonding is ionic and covalent, and which are generally formed at high temperatures.

Glass: (i) An inorganic product of fusion that has cooled to a rigid condition without crystallization; (ii) An amorphous solid.

Various microstructures



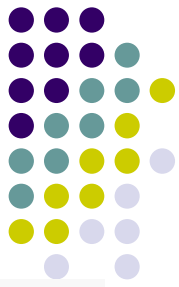
Amorphous: (i) Lacking detectable crystallinity; (ii) possessing only short-range atomic order; also *glassy* or *vitreous*

Bioactive material: A material that elicits a specific biological response at the interface of the material, resulting in the formation of a bond between the tissues and the material.

Crystalline versus Glassy Ceramics



- Crystalline ceramics have long-range order, with components composed of many individually oriented grains.
- Glassy materials possess only short-range order, and generally do not form individual grains.
- The distinction is based on x-ray diffraction characteristics.
- Most of the structural ceramics are crystalline and oxides.



Metal- Ceramic Comparison

Property	Units	Ti 6Al 4V	316 SS	CoCr Alloy	TZP	Alumina
Young's modulus	GPa	110	200	230	210	380
Strength	MPa	800	650	700	900-1200	> 500
Hardness	HV	100	190	300	1200	2200

- Stiffness is comparable to the metal alloys
- The biggest problem is fracture toughness (sensitivity to flaws).
- Rigid plastics < Metals < ceramics
- metals are ductile, whereas ceramics are brittle



Advantages:

- inert in body (or bioactive in body); Chemically inert in many environments
- high wear resistance (orthopedic & dental applications)
- high modulus (stiffness) & compressive strength
- esthetic for dental applications



Disadvantages

- brittle (low fracture resistance, flaw tolerance)
- low tensile strength (fibers are exception)
- poor fatigue resistance (relates to flaw tolerance)

Basic Applications:



Orthopedics:

- bone plates and screws
- total & partial hip components (femoral head)
- coatings (of metal prostheses) for controlled implant/tissue interfacial response
- space filling of diseased bone
- vertebral prostheses, vertebra spacers, iliac crest prostheses



Dentistry:

- dental restorations (crown and bridge)
- implant applications (implants, implant coatings, ridge maintenance)
- orthodontics (brackets)
- glass ionomer cements and adhesives

Veneers



Before and after





Other:

- inner ear implants (cochlear implants)
- drug delivery devices
- ocular implants
- heart valves

Glass-Ceramic cochlear implants



Ceramics



- Alumina, Zirconium, Hydroxyapatite, Calcium phosphates.
- Porous ceramic materials exhibit much lower strengths but have been found extremely useful as coatings for metallic implants.
- The coating aids in tissue fixation of the implant by providing a porous surface for the surrounding tissue to grow into and mechanically interlock.
- Certain ceramics are considered bioactive ceramics if they establish bonds with bone tissue.

Types of Bioceramic-Tissue Interactions:



Dense, inert, nonporous ceramics attach to bone (or tissue) growth into surface irregularities by press fitting into a defect as a type of adhesive bond (termed “morphological fixation”)- Al_2O_3

Porous inert ceramics attach by bone resulting from ingrowth (into pores) resulting in mechanical attachment of bone to material (termed “biological fixation”)- Al_2O_3

Dense, nonporous surface-reactive ceramics attach directly by chemical bonding with bone (termed “bioactive fixation”)-bioactive glasses & Hydroxyapatite.

Processing of Ceramics



1. Compounding

- Mix and homogenize ingredients into a water based suspension = slurry
or, into a solid plastic material containing water called a clay

2. Forming

- The clay or slurry is made into parts by pressing into mold (sintering). The fine particulates are often fine grained crystals.

3. Drying

- The formed object is dried, usually at room temperature to the so-called "green" or leathery state.

4. Firing

- Heat in furnace to drive off remaining water. Typically produces shrinkage, so producing parts that must have tight mechanical tolerance requires care.
- Porous parts are formed by adding a second phase that decomposes at high temperatures forming the porous structure.

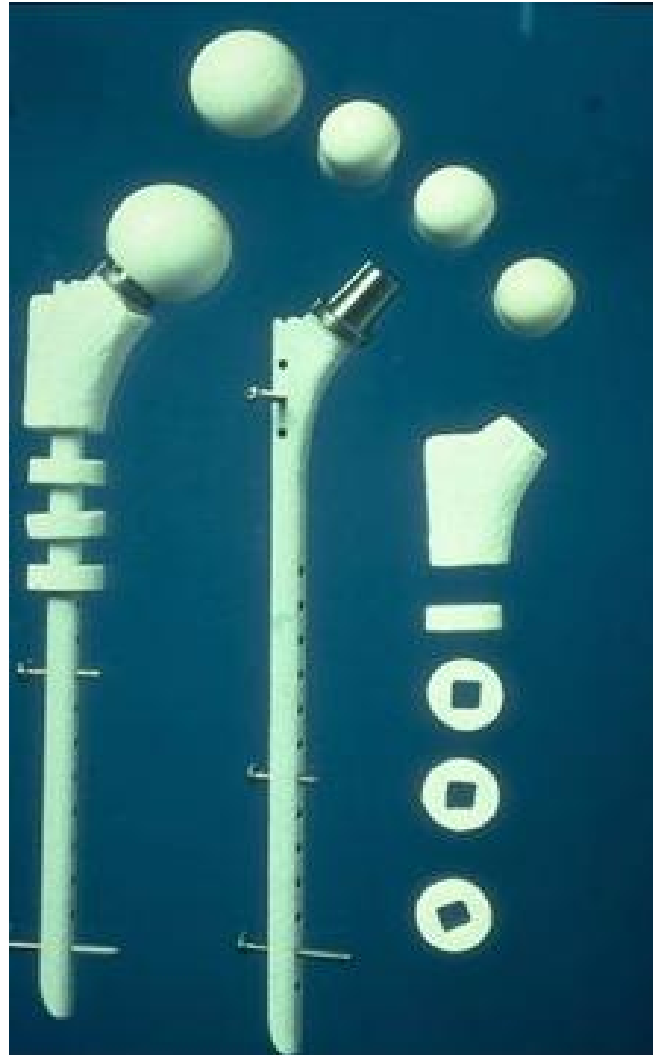
Alumina (Al_2O_3) and Zirconia (ZrO_2)

The two most commonly used structural bioceramics.

- Primarily used as modular heads on femoral stem hip components.
- wear less than metal components, and the wear particles are generally better tolerated.



Femoral Component



Alumina (Al_2O_3):

- single crystal alumina referred to as “Sapphire”
- “Ruby” is alumina with about 1% of Al^{3+} replaced by Cr^{3+} ; yields red color
- “Blue sapphire” is alumina with impurities of Fe and Ti; various shades of blue



Structure and Properties:



- most widely used form is polycrystalline
- unique, complex crystal structure
- strength increases with decreasing grain size
- elastic modulus (E) = 360-380 GPa

Fabrication of Biomedical devices from Al_2O_3 & (ZrO_2) :



- devices are produced by pressing and sintering fine powders at temperatures between 1600 to 1700°C.
- Additives such as MgO added (<0.5%) to limit grain growth



Dental Porcelain:

- ternary Composition = Mixture of $K_2O-Al_2O_3-SiO_2$ made by mixing *clays, feldspars, and quartz*

CLAY = Hydrated alumino silicate

FELDSPAR = Anhydrous alumino silicate

QUARTZ = Anhydrous Silicate



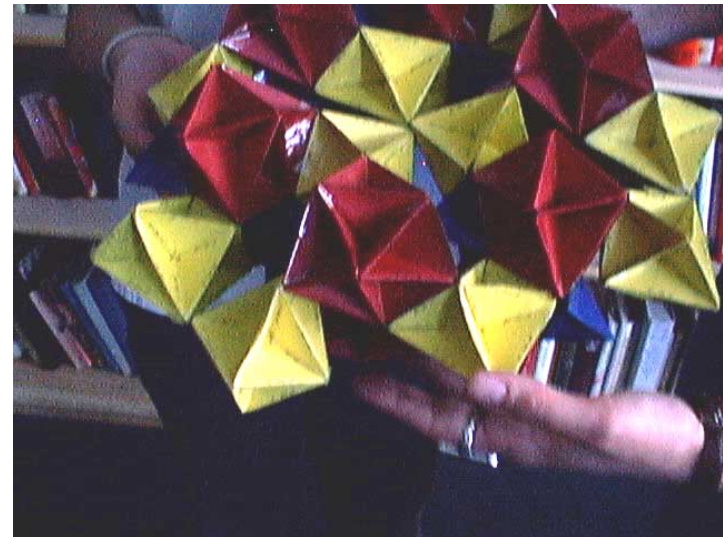
Calcium Phosphates

- Calcium phosphate compounds are abundant in nature and in living systems.
- Biologic apatites which constitute the principal inorganic phase in normal calcified tissues (e.g., enamel, dentin, bone) are carbonate hydroxyapatite, CHA.
- In some pathological calcifications (e.g., urinary stones, dental tartar or calculus, calcified soft tissues – heart, lung, joint cartilage)

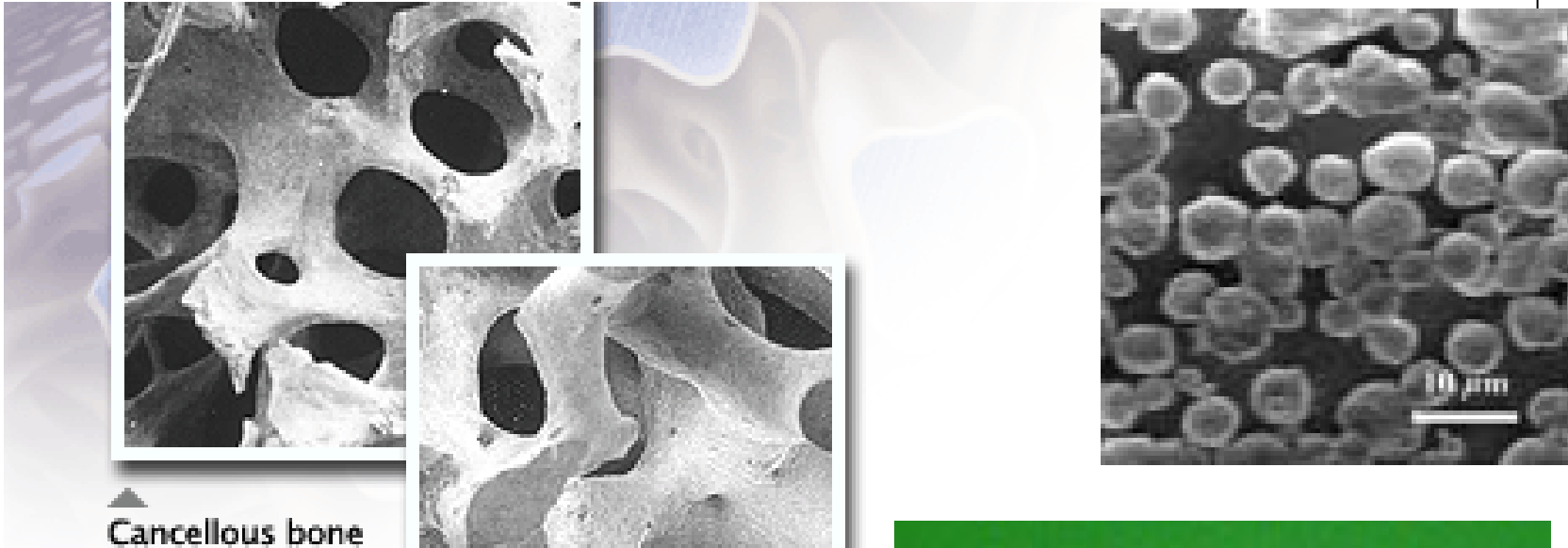
Calcium hydroxyapatite ($\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$): HA



- Hydroxyapatite is the primary structural component of bone. As its formula suggests, it consists of Ca^{2+} ions surrounded by PO_4^{2-} and OH^- ions.



Calcium hydroxyapatite ($\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$): HA

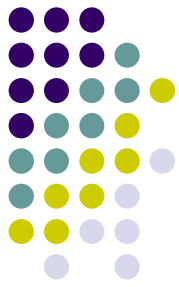


Calcium hydroxyapatite ($\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$): HA



- gained acceptance as bone substitute
- repair of bony defects, repair of periodontal defects, maintenance or augmentation of alveolar ridge, ear implant, eye implant, spine fusion, adjuvant to uncoated implants.

HA is :



- Since collagen is closely associated with HA in normal bone, it is a logical candidate for induction of a host response. In some cases bone growth in or near implanted HA is more rapid than what is found with control implants. In the literature HA is sometimes referred to as an "osteoinductive" material. However, HA does not seem to induce bone growth in the same way as, say, BMP.



Bioceramic Coatings

- Coatings of hydroxyapatite are often applied to metallic implants (most commonly titanium/titanium alloys and stainless steels) to alter the surface properties.
- In this manner the body sees hydroxyapatite-type material which it appears more willing to accept.
- Without the coating the body would see a foreign body and work in such a way as to isolate it from surrounding tissues.
- To date, the only commercially accepted method of applying hydroxyapatite coatings to metallic implants is plasma spraying.



Bone Fillers

- Hydroxyapatite may be employed in forms such as powders, porous blocks or beads to fill bone defects or voids.
- These may arise when large sections of bone have had to be removed (e.g. bone cancers) or when bone augmentations are required (e.g maxillofacial reconstructions or dental applications).
- The bone filler will provide a scaffold and encourage the rapid filling of the void by naturally forming bone and provides an alternative to bone grafts.
- It will also become part of the bone structure and will reduce healing times compared to the situation, if no